

Enhancing CO₂ Mineralisation in Steel Slag with Amines for Developing a Waste to Construction Material

*Takebe¹ K, Elakneswaran² Y, Yoda³ Y and Kitagaki⁴ R

¹Division of Cooperative Program for Resources Engineering, Graduate School of Engineering, Hokkaido University

²Division of Sustainable Resources Engineering, School of Engineering, Hokkaido University

³Shimizu Corporation, Shimizu Institute of Technology

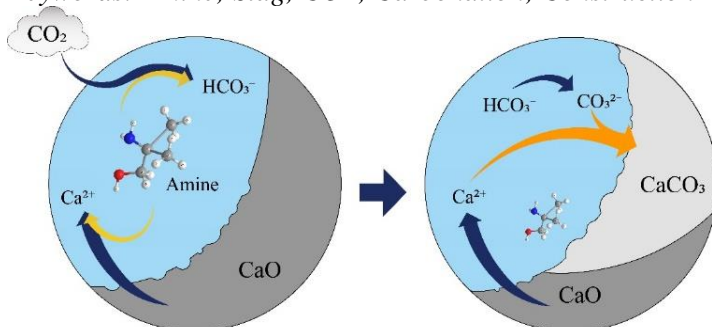
⁴Division of Architecture, Hokkaido University

*Corresponding author – Email: takebe.kanta.e3@elms.hokudai.ac.jp

Abstract

Global warming has become an urgent issue due to the increasing atmospheric emission of CO₂. In Japan, the steel industry emits a large amount of CO₂ and produces steel slag as an alkaline waste product. Using this steel slag to absorb CO₂ in flue gas is an effective approach for CO₂ emissions and recycling. This study introduces a novel technology utilizing amines to facilitate CO₂ sequestration in blast furnace slag. This innovative approach demonstrates significant potential and flexibility compared to conventional CO₂ capture technologies. In this study, changes in the carbonation efficiency of slag with three types of amines (N-Methyldiethanolamine (MDEA), 2-(Methylamino)ethanol (MAE), 2-Amino-2-methyl-1-propanol (AMP)) were investigated. In addition, this study examined the effect of Portland cement on the carbonation efficiency. In wet carbonation, the CO₂ fixation process entails the introduction of 1 mol/L of amine, water, slag, and cement, followed by a 24-hour leaching process and subsequent 24-hour carbonation. The results showed that adding AMP and cement exhibited the most significant increase in carbonation efficiency, resulting in the formation of 11.07% of the solid weight as CaCO₃, as confirmed by thermogravimetric analysis (TGA). Changes in calcium ion concentration and pH were also investigated in this study. The results showed that a small amount of Ca²⁺ dissolution and an increase in pH occurred in the early stages of the reaction and that the majority of Ca²⁺ dissolution occurred simultaneously with carbonation. In dry carbonation, slag cement paste, made by mixing amine, slag, and cement, was carbonated for 28 days after a 28-day curing period to determine changes in its properties. W/S ratio is 10, slag/cement ratio is 19, amine concentration is 1 mol/L, and CO₂ concentration is 0%, 0.04%, 5%, and 15%. Strength tests showed an increase in strength in the slag cement paste with the addition of MDEA.

Keywords: Amine; Slag; CO₂; Carbonation; Construction materials



This page is intentionally left blank

We hope these proceedings inspire further exploration and innovation